

Corrigenda

Electrical Conduction Properties of the Platinum Chain Complexes $(\text{NH}_4)_2\text{Pt}(\text{CN})_4\text{Cl}_{0.3}\cdot 3\cdot 2\text{H}_2\text{O}$ and $(\text{NH}_4)_2\text{Pt}(\text{CN})_4\text{Cl}_{0.5}\cdot 3\text{H}_2\text{O}$

By ALLAN E. UNDERHILL, DAVID M. WATKINS, and DAVID J. WOOD

J.C.S. Chem. Comm., 1976, 805.

Subsequent to the original work we have confirmed the existence of $(\text{NH}_4)_2\text{Pt}(\text{CN})_4\text{Cl}_{0.3}\cdot 3\cdot 2\text{H}_2\text{O}$ but have not been able to repeat the preparation of a compound having an NH_4^+ to Pt ratio of 2.2 : 1 even from solutions saturated in NH_4Cl . We must therefore conclude that our original sample may have been contaminated with NH_4Cl . We are investigating further the factors which give rise to variations in the analyses and in the conduction properties of the products of these preparations.

The Ramirez Dioxaphospholen Condensation: New Access to Branched-chain Sugars

By SERGE DAVID, MARIE-CHRISTINE LÉPINE, GUSTAVE ARANDA, and GEORGE VASS

J.C.S. Chem. Comm., 1976, 747.

The name of the third author should read: GÉRARD ARANDA; address: Ecole Polytechnique, 91120 Plateau de Palaiseau, France.

 $(\eta^4\text{-Cyclo-octa-1,4-diene})(\eta^6\text{-cyclo-octa-1,3,5-triene})\text{ruthenium}(0)$ Chemistry: the Role of Molecular Hydrogen in a New Synthetic Route to Cyclo-olefin Ruthenium Complexes

By PAOLO PERTICI, GIANPAOLO SIMONELLI, GIOVANNI VITULLI, GIULIO DEGANELLO, PIER LUIGI SANDRINI, and ANGELO MANTOVANI

J.C.S. Chem. Comm., 1977, 132.

The first word in the title should read: $(\eta^4\text{-Cyclo-octa-1,5-diene})(\eta^6\text{-cyclo-octa-1,3,5-triene})\text{ruthenium}(0)$.
The last author's name should read: ANTONIO MANTOVANI.

Configurational Rigid Tetrahedral Nickel Complexes used for Conformational Analysis of Cycloalkene Units

By RUDOLF KNORR* and ALFONS WEISS

J.C.S. Chem. Comm., 1977, 173.

Owing to a regrettable numerical error in conversion to SI units, the five energy values in this paper are too small by a common factor. The limit in the *Summary* should be 88 kJ mol^{-1} (21 kcal mol^{-1}). The correct numbers for the second paragraph are: $\geq 87.9 \text{ kJ mol}^{-1}$ for (1), ≥ 92.1 for (2), ≥ 94.6 for (3), > 87.5 for (4), and > 77.0 for (5). All other statements and the conclusions remain unchanged.

Reaction of Patchouli Alcohol with Lead Tetra-acetate; a New, Regiospecific Fragmentation of Patchouli Alcohol

By ALAN F. THOMAS and MICHEL OZAINNE

J.C.S. Chem Comm., 1977, 120.

In this communication, compound (1) was given the methanonaphthalene numbering. The *ab eo*-guaiane numbering for this system is as follows:

